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The Double [3 + 2] Photocycloaddition Reaction

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A common challenge confronting synthetic organic chemists is the atom-efficient creation of molecular complexity.¹ The meta photocycloaddition reaction² is useful in this context, as three new σ bonds, three new rings, and up to six new stereocenters are formed in a single step. More complexity can be added by tethering the two reacting partners together,^{3,4} and the elegant work of Wender^{2c-j} pioneered the reaction's application in natural product synthesis. While investigating the photochemistry of aromatic acetal **1**, we discovered a unique double [3 + 2] photocycloaddition reaction that resulted in the formation of the fenestrane⁵ derivative **2**⁶ (Scheme 1). This dramatic transformation represents the creation of five new rings, four new carbon–carbon σ bonds, and seven new stereocenters in a one-pot process.

Scheme 1. The Double [3 + 2] Photocycloaddition Reaction To Form Fenestrane $\mathbf{2}^{6\ a}$



^a Conditions: (a) hv (254 nm), cyclohexane, 18 h, 8%.

Although the synthetic utility of fenestrane-type structures has yet to be fully explored, a recent synthesis of the insecticide (–)-penifulvin A^{5d} might suggest an untapped potential. Theoretical interest in fenestranes remains high because of the distortion of the central quaternary carbon away from the standard tetrahedral bond angle of 109.5°. In the case of compound **2**, the transverse bond angle (C7–C8–C9) is 120.2° and the longitudinal bond angle (C12–C8–C1) is 128.5°.

Compound 1 was prepared in 98% yield using Noyori's acetalforming procedure⁷ by the reaction of allyloxytrimethylsilane with *o*-anisaldehyde in the presence of trimethylsilyl triflate. Irradiation of 1 in cyclohexane using 254 nm UV light for 18 h led to the formation of a multitude of products, including the pentacycle 2 in 8% yield (Scheme 1). In a separate experiment, 1 was irradiated to the point of its total consumption (4 h) to yield four major components: linear meta photocycloadduct 3, angular meta photocycloadduct 4, and two ortho-derived cycloadducts^{3d,8} **5a** and **5b** whose relative configuration could not be determined (Figure 1). The absence of any fenestrane adduct 2 formation under these conditions led us to conclude that it must have been generated sequentially from one of the monocyclized compounds shown in Figure 1.

To investigate this, a solution containing only the major meta photoadduct **3** in cyclohexane was irradiated using 254 nm UV light for 16 h; this resulted in the formation of fenestrane **2** in 38% yield (Scheme 2). We also discovered that small amounts of the angular meta photoadduct **4** and a rearranged photoproduct **6** were



Figure 1. The four principal photoadducts formed by irradiation of aromatic acetal **1** in cyclohexane for 4 h using 254 nm UV light.

also formed during this secondary irradiation process. It should be noted that compound 6 was difficult to obtain free of other contaminants because of coelution with 3.

 $\mbox{Scheme 2.}$ Conversion of Linear Meta Photoadduct 3 into Fenestrane $\mbox{2}^a$



^a Conditions: (a) hv (254 nm), cyclohexane, 16 h, 38%.

The minor meta photoadduct 4 could also be converted into 2 under the same photolytic conditions, although it primarily led to the formation of photoproduct 6.

To aid in obtaining a mechanistic understanding of these transformations and interpreting the NMR spectra, we prepared 1_D , a deuterated version of the aromatic starting material, from 5-bromo-2-anisaldehyde (7) (Scheme 3).

Scheme 3. Formation and Photolysis of Deuterated Substrate 1_D^a



^{*a*} Conditions: (a) 2.5 equiv of Me₃SiOCH₂CH=CH₂, 0.01 equiv of Me₃SiOTf, -84 °C, CH₂Cl₂, 87%; (b) *n*-BuLi, -78 °C, THF then MeOD, 78%; (c) *hv* (254 nm), cyclohexane, 18 h, 14%; (d) *hv* (254 nm), cyclohexane, 2 days, 11%; (e) *hv* ("old lamp"), cyclohexane, 7 days, 16%.

After conversion of 7 to bromoacetal 8 via the Noyori procedure⁷ with allyloxytrimethylsilane, metal—halogen exchange was accomplished using *n*-butyllithium, and the resulting anion was quenched with MeOD to afford $\mathbf{1}_D$ with >80% deuterium incorporation. The deuterated linear meta photoadduct $\mathbf{3}_D^9$ was obtained following an initial photolysis step, and this was then converted into the deuterated products $\mathbf{2}_D$ and $\mathbf{6}_D$ in secondary photolysis steps. We found this process to be lamp-dependent, as our initial attempt

to irradiate 3_D using an old 16 W low-pressure mercury vapor lamp resulted in the formation of $6_{\rm D}$ as the principal product instead of $2_{\rm D}$. This extended irradiation period also caused the complete consumption of 3_D , which aided in the chromatographic isolation of **6**_D.

The presence of various triplet sensitizers (acetone, acetophenone, and benzophenone) during the irradiation of compound 3 enhanced its conversion into the fenestrane product 2. The efficiency of this process was sensitizer-dependent, with acetone forming significant quantities of 2 in a matter of hours, although it had the disadvantage of forming other impurities. Benzophenone appeared to inhibit the transformation, with comparatively little conversion of 3 to 2 even after several days of irradiation. Acetophenone presented a better compromise, although the subsequent purification process was hampered by coeluting impurities. During these sensitized reactions, a small amount of photoequilibration between 3 and 4 was observed, although the formation of the rearranged product 6 was completely suppressed. The addition of isoprene as a triplet quencher inhibited the formation of 2 from 3, providing further evidence that a triplet state is involved in this process.

A plausible mechanistic rationale to account for the formation of 2_D and 6_D is presented in Scheme 4. Linear and angular meta photoadducts $(3_D \text{ and } 4_D)$ are known to interconvert under photolytic conditions.¹⁰ This could occur either by homolytic fission of the cyclopropyl ring (see 9_D) or by a [1,3]-sigmatropic rearrangement. Photoinduced homolytic fission of the external cyclopropyl ring bond of the linear meta photoadduct 3_D may afford diradical 10_D , which could cyclize onto the terminal alkene to create two new five-membered rings and hence $2_{\rm D}$. Photoexcitation of the angular meta photoadduct 4_D followed by single electron transfer (SET) from the methoxy group's lone pair of electrons to the external cyclopropyl ring bond would result in the fragmentation of the threemembered ring to afford 11_{D} . The allylic radical of 11_{D} could combine homolytically with the oxygen radical cation to form pseudo-methylated epoxide 12_D , which would fragment to give the doubly allylic ether 6_D .

Scheme 4. Mechanistic Proposal for the Formation of 2_D and 6_D



Wender, Dore, and deLong¹¹ prepared a similar fenestrane compound by reaction of the linear meta photocycloadduct 14 derived from the bisallyloxy acetal of o-tolualdehyde (13) with an acetonitrile radical. We found that 14 could be converted to fenestrane 15 after many days of irradiation, but it could not be isolated free of other contaminants (Scheme 5). Use of acetophenone as a sensitizer improved the rapidity of this process, although impurities again hindered purification.

Scheme 5. Photoconversion of o-Tolualdehyde-Derived Meta Photoadduct 14 into Fenestrane Adduct 15²



 a Conditions: (a) $h\nu$ (254 nm), cyclohexane, 9.5 h, 24%; (b) $h\nu$ (254 nm), cyclohexane, 8 days, <10%.

In conclusion, a remarkable double [3 + 2] photocycloaddition reaction resulting in the formation of fenestrane 2 from aromatic diene 1 has been reported. During this process, four carbon-carbon bonds, five new rings, and seven new stereocenters are created. The photoreaction occurred in a sequential manner from linear meta photocycloadduct 3, while rearranged photoproduct 6 was derived from angular meta photocycloadduct 4.

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Supporting Information Available: Experimental details, charcterization data for all new compounds, and crystallographic data (CIF) for 2. This material is available free of charge via the Internet at http:// pubs.acs.org.

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